

What mass of iron can be obtained from 16 tonnes of iron oxide?

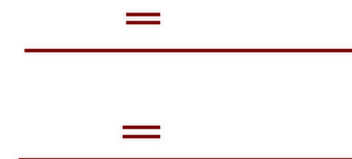
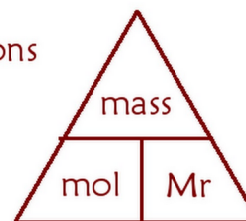
We are learning to:

- ① interconvert mass \rightleftharpoons mol using Mr
- ② use a balanced equation to work out amount (mol) of product formed

watch a video tutorial on this at www.chemistry.jamesmungall.co.uk

Definitions

①



②

Equation



Ratio

Mol

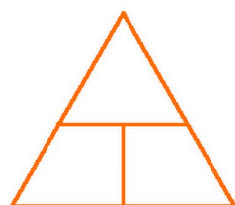
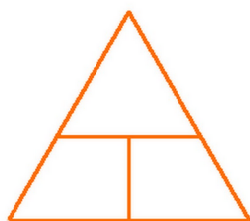
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(Fe = 56, O = 16)



Ratio _____

Mol _____



Summary

equation A \longrightarrow B

ratio :

mol

